

## Acid-Catalyzed Condensation of 1,3,5-Triphenylpentane-1,5-dione with Benzaldehyde—A New Domino Reaction

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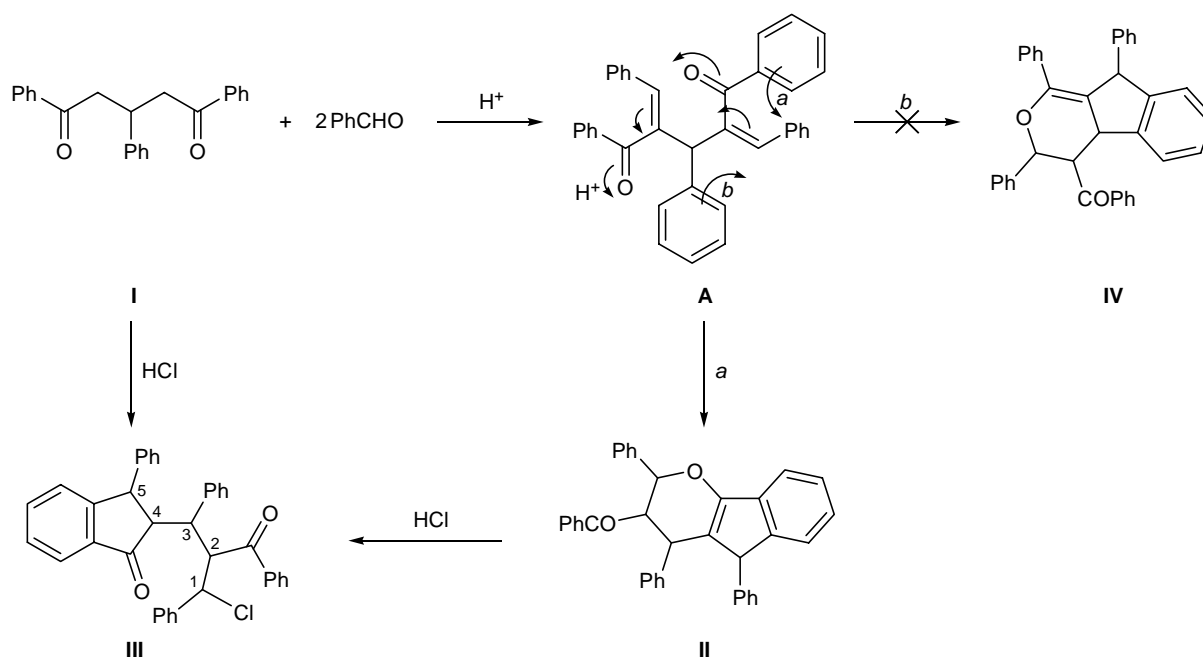
**Abstract**—Aldol-like condensation of 1,3,5-diphenyl-1,5-pentanedione with benzaldehyde in the presence of acids is accompanied by the Nazarov cyclization with formation of substituted indanone or indeno[1,2-*b*]pyran in a one-pot process.

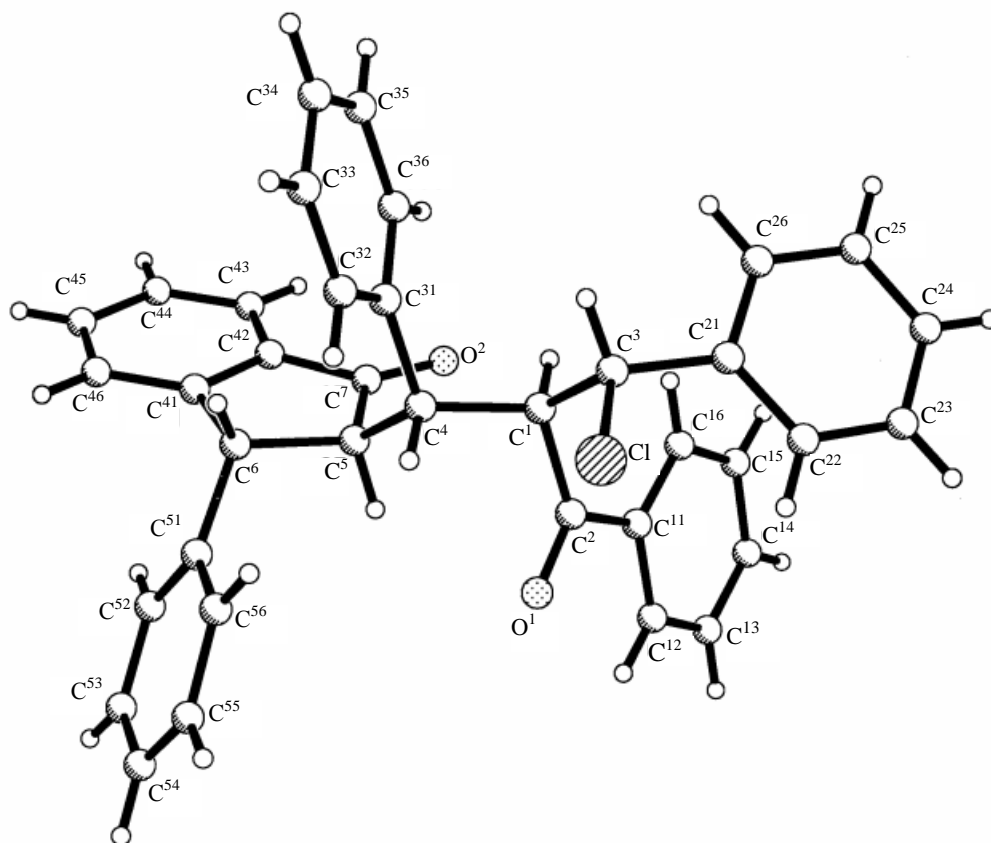
An actual problem of organic synthesis is search for new domino reactions [1] which make it possible to create complex structures in a one-pot process. While studying the reaction of 1,5-diketone **I** with benzaldehyde in the presence of sulfuric acid, we isolated compound **II**. Compound **II** was formed as a result of a new domino reaction which involves closure of three new carbon–carbon bonds via crotonization and Nazarov cyclization. When the reaction was initiated by gaseous hydrogen chloride, we obtained 2-(2-ben-

zoyl-3-chloro-1,3-diphenylpropyl)-3-phenylindan-1-one (**III**) which was also synthesized by treatment of **II** with hydrogen chloride. The structure of product **III** was proved by X-ray analysis [2] (see figure), thus confirming reaction path *a* and excluding alternative path *b* which would lead to structure **IV** (Scheme 1).

A product structurally related to **II**, 3-benzoyl-4-phenyl-2,3,4,5-tetrahydroindeno[1,2-*b*]pyran, was obtained previously by cyclization of 2,4-dibenzoyl-3-phenyl-1,4-pentadiene [3].

Scheme 1.





Structure of 2-(2-benzoyl-3-chloro-1,3-diphenylpropyl)-3-phenylindan-1-one according to the X-ray diffraction data.

## EXPERIMENTAL

The IR spectrum was recorded on a Perkin–Elmer Spectrum BX FT-IR System spectrometer. The mass spectrum (chemical ionization) was obtained on an Agilent Series 1100 LC/MSD instrument. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were measured on an Avance 500 spectrometer (500 MHz for  $^1\text{H}$ ). X-Ray diffraction data were obtained with the aid of a SMART-1000 CCD diffractometer ( $\text{MoK}_\alpha$  radiation, graphite monochromator).

**3-Benzoyl-2,4,5-triphenyl-2,3,4,5-tetrahydroindeno[1,2-*b*]pyran (II).** Concentrated sulfuric acid, 35 g, was added over a period of 2.5 h under vigorous stirring to a mixture of 6.56 g (0.02 mol) of diketone **I** and 6.5 g (0.06 mol) of benzaldehyde, maintained at  $-15^\circ\text{C}$ . The mixture was kept for 2 days at  $-4$  to  $0^\circ\text{C}$ , and the crystalline product was separated, washed with water, and recrystallized from benzene. Yield 4.03 g (40%), mp  $191$ – $193^\circ\text{C}$ . IR spectrum ( $\text{CH}_2\text{Cl}_2$ ),  $\nu$ ,  $\text{cm}^{-1}$ : 1681 (C=O), 1652 (C=C–O).  $^1\text{H}$  NMR spectrum ( $\text{CDCl}_3$ )  $\delta$ , ppm: 3.85 d (1H, 4-H,  $J = 5.8$  Hz), 4.32 s (1H, 5-H), 4.52 d.d (1H, 3-H,  $J = 5.8, 10.6$  Hz), 5.67 d

(1H, 2-H,  $J = 10.6$  Hz); signals from aromatic protons are omitted.  $^{13}\text{C}$  NMR spectrum ( $\text{CDCl}_3$ ),  $\delta_{\text{C}}$ , ppm: 41.8 ( $\text{C}^4$ ), 52.1 ( $\text{C}^5$ ), 52.3 ( $\text{C}^3$ ), 76.8 ( $\text{C}^2$ ), 197.7 (C=O); signals from aromatic carbon nuclei are omitted. Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 505.2 [ $M + 1$ ] (100), 297.1 [ $M + 1 - \text{C}_6\text{H}_5\text{CH}=\text{CHCOC}_6\text{H}_5$ ] (37), 209.1 [ $\text{C}_6\text{H}_5\text{CH}=\text{CHCOC}_6\text{H}_5 + 1$ ] (10). Found, %: C 87.90; H 5.90.  $\text{C}_{37}\text{H}_{28}\text{O}_2$ . Calculated, %: C 88.06; H 5.59.

**2-(2-Benzoyl-3-chloro-1,3-diphenylpropyl)-3-phenylindan-1-one (III).** *a.* Dry hydrogen chloride was passed over a period of 3 h through a suspension of 4.9 g (0.015 mol) of diketone **I** in 7.2 g (0.07 mol) of benzaldehyde, maintained at  $-7^\circ\text{C}$ , and the mixture was kept for 3 days at  $-4$  to  $0^\circ\text{C}$ . The crystalline product was filtered off, washed with diethyl ether, and recrystallized from ethyl acetate. Yield 2.73 g (38%), mp  $199$ – $200^\circ\text{C}$ .

*b.* Dry hydrogen chloride was passed over a period of 6 h through a solution of 0.54 g (1 mmol) of compound **II** in 2 ml of a 1:10 (by volume) mixture of anhydrous ethanol and dioxane, maintained at  $-5^\circ\text{C}$ . The mixture was evaporated, and the residue was

recrystallized from ethyl acetate. Yield 0.34 g (64%), mp 199–200°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 1702, 1675 (C=O).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 3.07 t (1H, 4-H,  $J = 5.0, 5.0$  Hz), 4.03 d.d (1H, 3-H,  $J = 5.1, 11.1$  Hz), 4.05 d (1H, 5-H,  $J = 5.1$  Hz), 4.84 d (1H, 1-H,  $J = 3.2$  Hz), 5.72 d.d (1H, 2-H,  $J = 3.2, 11.1$  Hz).  $^{13}\text{C}$  NMR spectrum,  $\delta_{\text{C}}$ , ppm: 46.4 ( $\text{C}^3$ ), 49.6 ( $\text{C}^5$ ), 54.5 ( $\text{C}^2$ ), 56.6 ( $\text{C}^4$ ), 63.4 ( $\text{C}^1$ ), 199.5 (C=O), 208.0 (C=O). Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 541.5 [ $M + 1$ ] (100), 505 [ $M + 1 - \text{Cl}$ ] (78.5), 297.1 [ $M + 1 - \text{C}_6\text{H}_5\text{CH}=\text{CH}-$

$\text{COC}_6\text{H}_5$ ] (29.7), 209.1 [ $\text{C}_6\text{H}_5\text{CH}=\text{CHCOC}_6\text{H}_5 + 1$ ] (5). Found, %: Cl 6.59.  $\text{C}_{37}\text{H}_{29}\text{ClO}_2$ . Calculated, %: Cl 6.56.

## REFERENCES

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3. Pavel', K.G., Pavel', G.V., Vysotskii, V.I., and Tilichenko, M.N., *Zh. Org. Khim.*, 1991, vol. 27, p. 1607.